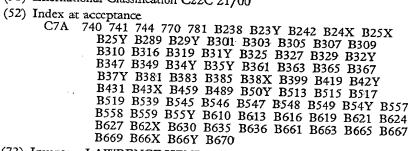
PATENT SPECIFICATION

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(54) IMPROVEMENTS IN OR RELATING TO ALUMINIUM ALLOYS

We, UNITED KINGDOM ATOMIC ENERGY AUTHORITY, London, a British Authority, do hereby declare the invention, for which we pray that a patent may be granted to us, and the method by which it is to be performed, to be particularly described in and by the following state-

This invention relates to wrought aluminium alloys, and in particular to aluminium based alloys containing copper, magnesium, silicon, manganese and incidental impurities.

According to the invention aluminium

based alloys contain copper 3.5-5.0 wt %, magnesium 0.2—1.0 wt %, silicon 0.5—1.0 wt %, manganese 0.3—1.5 wt %, silver 0.1— 1.0 wt %, remainder aluminium with incidental impurities including iron up to 0.7 wt % maximum.

A more restricted range of alloys according to the invention is the range represented by the specification: copper 4.0-4.7 wt %, magnesium 0.45—0.8 wt %, silicon 0.7—0.9 wt %, manganese 0.6—0.9 wt %, silver 0.2—0.7 wt %, remainder aluminium with incidental impurities including iron up to 0.5 wt % maximum.

Alloys according to the invention when given conventional heat treatment, for example solution treated at 500°C. for 16 hours, cold water quenched followed by aging at 160°C. for 30 hours have been shown to possess high room temperature tensile properties which are improved as compared with the room temperature tensile properties of the silver free alloys when similarly heat treated.

Typically alloys in accordance with the invention but with zero silver content possess

an average room temperature 0.1% proof stress averaging 27.5 tons/in2, an average ultimate tensile stress (UTS) averaging 31.5 tons/in2 and a percentage elongation of about 10%. By way of comparison alloys in accordance with the invention have been found to possess a room temperature 0.1% proof stress generally in the range 27.8—29 tons/in², a UTS in the range 31.8—32.8 tons/in2 whilst retaining a percentage elongation at least as good as that of the silver free alloys. It has been found that in alloys in accordance with the invention such improved room temperature tensile properties are largely maintained at elevated temperatures in the range 80-

Alloys in accordance with the invention have also been found to possess an improved creep resistance at elevated temperatures as compared with the silver free alloys. Having established improved creep resistance at elevated temperatures, it is a logical expectation that improved creep resistance at room temperature will be experienced.

In the case of the silver free alloys it has been found that a long term soaking in a condition of no applied stress at a temperature in the range 70-150°C. for a period of from 100 to 1000 hours results in an improvement in the room temperature and elevated temperature tensile properties and in the room temperature creep resistance of such alloys. It has also been found that a long term soaking as aforesaid is similarly advantageous with respect to the silver containing alloys in accordance with the invention.

Examples of alloys in accordance with the invention and containing silver additions in



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the range 0.1—1.0 percent are given below. Examples (b) and (g) are slightly outside the said silver range. Test results are also quoted showing the effect of such silver additions in giving an improvement in the room temperature and elevated temperature tensile propertics of the alloys, in giving an improvement in the room temperature and elevated temperature creep resistance of the alloys. The later results also show the effect of a long term soaking in a condition of no applied stress in giving a firmer improvement in the room temperature tensile properties, elevated

temperature tensile properties and room tem-

perature creep resistance of the alloys. In each case tests in respect of silver free alloys are given for comparison.

The addition of silver in the range 0.1-1.0 wt % improves the room temperature tensile properties

Wrought form of test specimens: Forged Heat Treatment

Solution treatment by heating for 16 hours at 500°C. followed by a cold water quench and precipitation hardening by heating for 20 hours at 170°C.

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Room Temperature Tensile Tests

			Composi	tion							
	Cu	Mg	Si	Mn	Ag	Fe	0.1° o PS (Tons/in ²	UTS) (Tons/in²)	o EL		
(a)	4.21	0.71	0.76	0.77	_	0.13	27.8	32.0	11		
(b)	4.14	0.69	0.79	0.80	0.09	0.15	27.9	31.8	12		
(c)	4.17	0.70	0.81	0.82	0.19	0.15	27.9	32.1	12		
(d)	4.11	0.71	0.82	0.81	0.28	0.16	27.9	32.1	11		
(e)	4.10	0.71	0.82	0.81	0.38	0.16	27.8	32.0	10		
(f)	4.19	0.71	0.83	0.81	0.48	0.15	28.8	32.8	11		
(g)	4.27	0.71	0.81	0.75	1.02	0.15	29.0	33.4	9		
(h)	4.31	0.47	0.79	0.77	. ••	0.12	27.4	31.3	.11 .		
(i)	4.16	0.46	0.78	0.80	0.40	0.12	28.0	32.1	11		

In the above table example (a) is given for comparison and relates to a silver free alloy. Examples (c) to (f) demonstrate the effect of the addition of increasing silver content in the range 0.1—1.0 wt %, Examples (b) and (g) having silver just outside this range.

Example (h) is also given for comparison and relates to a silver free alloy having a

different base composition to that of example (a). Example (i) demonstrates the effect of an 0.4 wt % silver addition to the silver free alloy of example (h).

The addition of silver improves the room temperature tensile properties of alloys having a range of compositions Wrought form of test specimens: Examples (a)—(j) Forged Examples (k)—(n) Extruded

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Heat Treatment: Solution treatment by heating for 16 hours at 500°C. followed by a cold water quench and precipitation hardening by heating for 20 hours at 170°C.

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		Co	ompositio	n			Room Temperature Tensile Tests				
	Cu	Mg	Si	Mn	Ag	Fe	0.1% PS (Tons/in ²	UTS) (Tons/in²)	% EL		
(a)	4.31	0.47	0.79	0.77		0.12	27.4	31.3	11		
(b)	4.21	0.71	0.76	0.77		0.13	27.8	32.0	10		
(c)	3.90	0.70	0.83	0.80	0.38	0.13	27.8	32.0	10		
(d)	4.56	0.70	0.80	0.80	0.38	0.12	28.3	32.6	11		
(e)	4.16	0.46	0.78	0.80	0.40	0.12	27.6	31.8	11		
(f)	4.20	0.79	0.79	0.73	0.39	0.14	28.6	32.8	10		
(g)	4.24	0.71	0.59	0.79	0.40	0.12	27.9	32.1	11		
(h)	4.31	0.71	0.92	0.78	0.39	0.13	28.2	32.4	10		
(i)	4.18	0.71	0.79	0.60	0.39	0.13	28.4	32.6	11		
(j)	4.22	0.71	0.80	0.91	0.39	0.14	28.6	32.8	10		
Extrude	ed 										
(k)	4.02	0.47	0.72	0.64		0.11	26.3	30.7	7		
(1)	3.86	0.45	0.72	0.64	0.21	0.11	27.5	32.4	10		
(m)	4.67	0.81	0.93	0.95		0.12	26.5	31.5	9		
(n)	4.33	0.66	0.92	0.86	0.69	0.14	31.3	35	11		

In table (2) above examples (a) and (b) are given for comparison and relate to silver free alloys. Examples (c) to (j) demonstrate the improving effect of approximately 0.4 wt % silver with variation of amounts of all the major elements. In table (2) examples (b) % silver with variation or amounts or all the major elements. In table (2) examples (k) and (m) are also given for comparison and relate to silver free alloys whereas examples (l) and (n) respectively contain 0.21 and 0.69 wt % silver. Examples (k) to (n) show the effect of variation of all the major elements

including silver between the minima and maxima.

3) The addition of silver improves the room temperature tensile properties of alloys having a wide range of compositions Wrought form of test specimens: Forged Heat Treatment

Solution treatment by heating for 16 hours at 500°C. followed by a cold water quench and precipitation hardening by heating for 20 hours at 170°C.

			Co	mpositior	1		Room Temp	erature Ten	sile Tests
	Cu	Mg	Si	Mn	Ag	Fe	0.1% PS (Tons/in²)	UTS (Tons/in²)	% EL
(a)	4.31	0.47	0.79	0.77	_	0.12	27.4	31.3	11
(b)	4.21	0.71	0.76	0.77		0.13	27.8	32.3	10
(c)	3.4	0.71	0.80	0.76	0.42	0.16	27.1	31.6	12
(d)	4.98	0.71	0.82	0.77	0.42	0.15	28.7	32.3	8
(e)	4.22	0.20	0.80	0.77	0.42	0.15	25.0	29.9	12
(f)	4.33	1.03	0.81	0.76	0.42	0.10	28.1	32.2	11
(g)	4.16	0.42	0.80	0.79	0.49	0.14	29.0	32.4	10.5

In table (3) above examples (a) and (b) are given for comparison and relate to silver free alloys. Examples (c) to (g) demonstrate that the improvement in room temperature tensile properties caused by silver additions of about 0.4 wt % is effective over a wide range of base composition. Example (c)—low copper, and example (f)—high magnesium, are slightly outside the ambit of the invention. The results are mainly for wide variations in copper and magnesium which are the two

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elements most likely to have a significant effect on the mechanical properties.

4) The addition of silver is effective in different wrought forms Heat Treatment

Solution treatment by heating for 16 hours at 500°C. followed by a cold water quench and precipitation hardening by heating for 20

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Wrought form of test specimens: Forged

			Compo	ition			Room Tem	perature Ten	sile Tests
	Cu	Mg	Compos	Mn	Ag	Fe Fe	0.1% PS (Tons/in²)		% E1.
(a)	4.21		0.76			0.13	27.9	31.4	70
(b)	4.10	0.71	0.82	0.81	0.38	0.16	28.5	32.5	8

hours at 170°C.

Wrought form of test specimens: Extruded Bar

			Comp	ocition			Room Tem	perature Ten	sile Tests
	Cu	Mg	Si	Mn	Ag	Fe	0.1% PS (Tons/in²)		% EL
(c)	4.21	0.71	0.76	0.77		0.13	31.2	34.2	10
(d)	4.23	0.72	0.79	0.75	0.39	0.13	32.4	35.2	10

In table (4) above examples (a) and (c) are given for comparison and relate to silver free alloys. The examples (b) and (d) show that the improving effect of silver additions on the mechanical properties of alloys is not dependent on the wrought form of the alloy.

5) The addition of silver in the range 0.1— 1.0 wt % improves the elevated temperature tensile properties Heat Treatment:

Solution treatment by heating for 16 hours at 500°C. followed by a cold water quench and precipitation hardening by heating for 20 hours at 170°C.

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										田	levated Ter	nperatur	Elevated Temperature Tensile Tests at:	ests at:		
					٠			2°08			J00°C			120°C		
	Ö	Mg	Si	Mn	Ag	Fe	0.1% P.S. (tons/in²)	U.T.S. (tons/in²)	% EI	0.1% P.S. (tons/in²)	U.T.S. (tons/in²)	% EI	0.1% P.S. (tons/in²)	U.T.S. (tons/in²)	% EI	
(a)	4.21	0.71	0.76	0.77	1	0.13	26.8	29.6	12	27.1	29.6	102	26.4	28.2	- 2	
(b)	4.14	0.69	0.79	08.0	0.09	0.15							0 90	1 0 2 0	2 2	
<u> </u>	4.17	0.70	0.81	0.81	0.19	0.15							25.5	2. 6	G 5	_
(d)	4.11	0.71	0.82	0.81	0.28	0.16	28.0	30.8	=	27.5	30.2	Ξ	7.7.2	6.12	7 9	
<u>ම</u>	4.10	0.71	0.82	0.81	0.38	0.16	28.3	30.6		27.4	. «	; ;	7:17	7.67	0, ;	
(f)	4.19	0.71	0.83	0.81	0.48	0.15				: :	2	07	4. 17	29.3		1
(g	4.27	0.71	0.81	0 75	1	0 15							27.0	29.4	4	1,320
9		•	5	3.	1.04	0.15							28.2	30.8	7,2	0,20
Wrou	ght form	of test s	Wrought form of test specimens: Extruded bar	: Extru	ded bar	·										71
(h)	4.21	0.71	0.76	0.77	I	0.13	29.6	32.1	17	٠			27.8	30 1	10	
(<u>:</u>)	4.23	0.72	0.79	0.75	0.39	0.13	30.8	33.2	18				8 80	30.02		
														9.5	2	
			Ir. are	table (5) above comparis	example son and	In table (5) above examples (a) and (h) are given for comparison and relate to silver		The addition sistance	6) The addition of silver improves creep resistance	inproves cr	-ə.ı dəə			·	
			1ree 5 (b) 2	alloys. 1. and (9) a	he remain	rictly ac	Iree alloys. The remaining examples (examples (b) and (g) are not strictly according to the		Heat Treatment:	ent:						
				ontion on	silver con	tent) show	invention on silver content) show the improve		Columbian turn	-	•	,			_	

invention on silver content) show the improving effect of silver additions on the mechanical properties at elevated temperatures and that the improvement is not critically dependent on the wrought form of the alloy.

15 Solution treatment by heating for 16 hours at 500°C. followed by a cold water quench and precipitation hardening by heating for 20 hours at 170°C.

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(f)

4.23

Wrought form of test specimens: Forged

				•.•.			To	tal Plastic	Strain (%)	at
	Cu	Mg	Compos	Mn	Ag	Fe	70°C/26 500 hours	tonf/in² 1000 hours	100°C/20 500 hours	
(a)	4.21	0.71	0:76	0.77		0.13	.380	.400	.028	.032
(b)	4.11	0.71	0.82	0.81	0.28	0.16	.124	.130	_	
(c)	4.10	0.71	0.82	0.81	0.38	0.16	.101	.105	.017	.018
(d)	4.27	0.71	0.81	0.75	1.02	0.15	·			
Wrought	form of t	test speci								
(e)	4.21	0.71	0.76	0.77		0.13	.275	.32		

0.39

0.13

In table (6) above examples (a) and (e) are given for comparison and relate to silver free alloys. The remaining examples (example 5 (d) is not strictly according to the invention on silver content) show the improving effect of silver additions on the creep resistance. The results also show that the effect is not dependent on the wrought form of the alloy.

0.72

0.79

0.75

7) Long term soaking in a condition of no applied stress over a range of temperatures improves the Room Temperature Tensile Properties of both silver free and silver containing alloys

Wrought form of test specimens: Forged Primary Heat Treatment

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Solution treatment by heating for 16 hours at 500°C. followed by a cold water quench and precipitation hardening by heating for 20 hours at 170°C.

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			Comp	osition			Soaking		Room Temperature Tensile Tests		
	Cu	Mg	Si	Mn	Ag	Fe	- Temp. °C	Time Hours	0.1% P. (Tons/i	S. U.T.S n ²) (Tons/i	n ²) % Fi
(a)	4.2	0.42	0.84	0.76	0.31	0.13			29.0	32.7	11
(b)	4.2	0.42	0.84	0.76	0.31	0.13	100	500	29.4	33.2	11
(c)	4.2	0.42	0.84	0.76	0.31	0.13	100	750	29.6	33.4	12
(d)	4.2	0.42	0.84	0.76	0.31	0.13	100	1000	29.8	33.2	10
(e)	4.2	0.42	0.84	0.76	0.31	0.13	120	250	29.8	33.4	10
(f)	4.2	0.42	0.84	0.76	0.31	0.13	120	500	28.8	33.4	13
(g)	4.2	0.42	0.84	0.76	0.31	0.13	120	750	29.6	33.9	11
(h)	4.2	0.42	0.84	0.76	0.31	0.13	120	1000	29.6	33.0	11
(i)	4.2	0.42	0.84	0.76	0.31	0.13	135	100	29.6	33.3	11
(j)	4.2	0.42	0.84	0.76	0.31	0.13	135	250	29.6	33.4	10
(k)	4.2	0.42	0.84	0.76	0.31	0.13	135	500	29.4	32.9	11
(1)	4.2	0.42	0.84	0.76	0.31	0.13	135	750	29.3	33.0	10
(m)	4.2	0.42	0.84	0.76	0.31	0.13	135	1000	28.7	33.4	13
							Second A	geing			13
(n)	4.21	0.71	0.76	0.77		0.10	Treatn	nent ———			
(o)	4.21	0.71	0.76	0.77		0.13			27.8	32.0	11
(p)	4.21	0.71	0.76	0.77		0.13	1000h/70		27.7	31.8	9
(g)	4.21	0.71	0.76	0.77		0.13	1000h/10		28.1	32.0	9
(r)	4.10	0.71	0.82	0.77		0.13	1000h/15	50°C	26.5	30.4	8
(s)	4.10	0.71	0.82		0.38	0.16			27.8	32.0	10
(t)	4.10	0.71	0.82	0.81 0.81	0.38	0.16	1000h/70		28.6	32.8	10
(u)	4.10	0.71	0.82		0.38	0.16	1000h/10		28.7	32.7	9
(v)	4.11	0.71	0.82	0.81	0.38	0.16	1000h/15	:0°С	27.5	31.5	9
(w)	4.11	0.71	0.82	0.81	0.28	0.16			27.9	32.1	11
(x)	4.31			0.81	0.28	0.16	1000h/10	0°C	29.2	33.2	10
(y)	4.31	0.47	0.79	0.77	_	0.12	_		27.4	31.3	11
(y) (z)		0.47	0.79	0.77	-	0.12	1000h/10	0°C	28.2	32.2	11
	4.21	0.71	0.76	0.77	-	0.13			27.8	32.0	11
(z¹)	4.21	0.71	0.76	0.77		0.13	1000h/10	0°C	28.1	32.0	9

In table (7) above examples (b) to (m) demonstrate the effect of an additional prolonged soaking in a condition of no applied stress at elevated temperature in improving the room temperature tensile properties of alloys containing 0.31 wt ½ silver. Example (a) relates for comparison to an alloy which has not been given an additional soaking treatment.

Examples (b) to (m) show that the effect depends both on the duration and temperature of the soaking. Within certain limits the lower the temperature of the soaking the longer is the time required for maximum enhancement of the mechanical properties. Conversely the higher the temperature of the soaking the shorter is the time required for maximum enhancement of the mechanical properties. Also maximum enhancement of mechanical properties is generally achieved by soaking at the lower temperatures for the longer times

Examples (0), (p) and (q) show the variation of the effect with temperature in soaking silver free alloys for 1000 hour periods. Example (n) relates for comparison to a silver free alloy of similar base composition to the alloys of examples (0) (p) and (q) but which has not been given an additional soaking treatoner.

Examples (s) (t) and (u) show the variation of the effect with temperature in soaking alloys containing 0.38 wt % silver for 1000

hour periods. Example (r) relates for comparison to an alloy of similar composition to the alloys of examples (s) (t) and (u) (i.e. also containing 0.38 wt % silver but which has not been given an additional soaking treatment).

Example (w) shows the effect of soaking an alloy containing 0.28 wt % silver for 1000 hours at 100°C, as compared with a similar alloy (example v) which has not been given an additional soaking treatment. Comparison of example (v) with example (s) (t) and (u) demonstrates that the effect applies to alloys of different silver content.

Examples (x) (y) (z) and (z^1) demonstrate that the effect applies to silver free alloys of different base compositions (i.e. containing 0.47 wt %, magnesium in examples (x) and (y) and containing 0.71 wt %, magnesium in examples (z) and (z^1).

8) Long term soaking in a condition of no applied stress at an elevated temperature (e.g. for 1000 hours at 100°C.) improves the Elevated Temperature Tensile Properties of both silver free alloys and alloys containing silver.

Wrought Form of Test Specimens: Forged Primarly Heat Treatment

Solution treatment by heating for 16 hours at 500°C. followed by a cold water quench and precipitation hardening by heating for 20 hours at 170°C.

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၁့၀	120°C % El		13	12	I	1	14	7
Tests at 12	U.T.S.	(10115/111-)	28.2	29.2	I	i	29.3	29.8
Elevated Temperature 100°C Tensile Tests at 120°C	0.1% P.S. 1	(mr/smo -)	26.4	27.4	J	i	27.4	27.8
re 100°C	% EI		10	13	ł	į	. 01	7
Temperatu	U.T.S.		29.6	29.5	1	1	29.8	31.1
Elevated	0.1% P.S. (Tons/in²) (27.1	27.2	l	1	27.4	28.6
	% EI		12	6	11	œ	11	æ
2°08	U.T.S. (Tons/in²)		29.6	30.2	30.8	31.4	3.06	31.8
	0.1% P.S. (Tons/in ²) (26.8	27.8	28.0	28.4	28.3	28.8
Second	Ageing Treatment		i	000h/100°C	I	1000h/100°C	ĺ	0.16 1000h/100°C 28.8
	Fe		0.13	0.13 1	0.16	0.16	0.16	0.16
Ç.	Ag		l	l	0.28	0.28	0.38	0.38
Composition	Mn		0.77	0.77	0.81	0.81	0.81	0.81
ပိ	Si	;	0.76	0.76	0.82	0.82	0.82	0.82
	Cu Mg Si Mn Ag		0.71	0.71	0.71	0.71	0.71	0.71
	ō	;	(a) 4.21 0.71 0.76 0.77	(b) 4.21 0.71 0.76 0.77	(c) 4.11 0.71 0.82 0.81 0.28	(d) 4.11 0.71 0.82 0.81 0.28	(e) 4.10 0.71 0.82 0.81 0.38	(f) 4.10 0.71 0.82 0.81 0.38
		;	(a)	(p)	(<u>c</u>)	(g)	©	Ξ

6 In table (8) above examples (a) (c) and (e) grelate to alloys of similar base composition and containing respectively no silver, 0.28 wt % silver, and 0.38 wt % silver, the alloys (a) (c) and (e) not having been given any additional soaking treatment. Alloys (b) (d) and (f) correspond to and are of similar composition to alloys (a) (c) and (e) but have been soaked in a condition of no applied stress a for 1000 hours at 100°C, and demonstrate the improvement in elevated temperature mechanical properties obtained by such a further 5 2

20 25 9) Long term soaking in a condition of no applied stress at an elevated temperature (e.g. for 1000 hours at 100°C.) improves the creep resistance of both silver free alloys and alloys containing silver.
Wrought form of test specimens: Forged Heat Treatment Solution treatment by heating for 16 hours at 500°C. followed by a cold water quench and precipitation hardening by heating for 20

hours at 170°C.

		Con	mposition	L		0 14-2		c Strain % at 5 tonf/in²
Cu	Mg	Si	Mn	Ag	Fe	Second Ageing Treatment	500 hours	1000 hours
4.21	0.71	0.76	0.77	_	0.13		.380	.400
4.21	0.71	0.76	0.77		0.13	1000h/100°C	. 140	.160
4.10	0.71	0.82	0.81	0.38	0.16	_	. 101	.105
4.10	0.71	0.82	0.81	0.38	0.16	1000h/100°C	.040	.045

Alloys according to the invention may also include minor improving elements such as one or more of titanium, lithium, beryllium, zirconium, vanadium, boron, cadmium and germanium. These may be present each up to 0.5 wt % maximum and in total not more than 1.0 wt %.

WHAT WE CLAIM IS: --

1. Aluminium based alloys containing 3.5—5.0 weight percent copper, 0.2—1.0 weight percent magnesium, 0.5—1.0 weight percent silicon, 0.3—1.5 weight percent manganese, 0.1—1.0 weight percent silver, remainder aluminium with incidental impurities including iron up to 0.7 weight percent maximum.

2. Aluminium based alloys containing 4.0—4.7 weight percent copper, 0.45—0.8 weight percent magnesium, 0.7—0.9 weight percent silicon, 0.6—0.9 weight percent manganese, 0.2—0.7 weight percent silver, remainder aluminium with incidental impurities including iron up to 0.5 weight percent maximum.

3. Aluminium alloys as claimed in claim 1 and substantially as hereinbefore described with reference to the examples given in the accompanying specification.

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